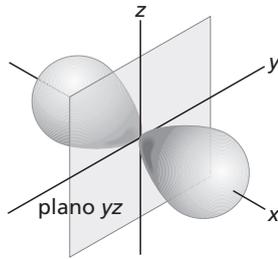


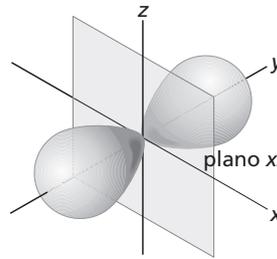
HIBRIDACIÓN de ORBITALES ATÓMICOS



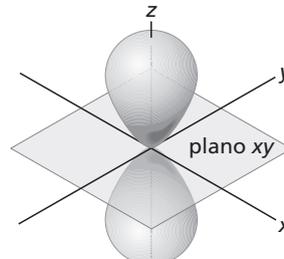
Orbital s.



orbital p_x

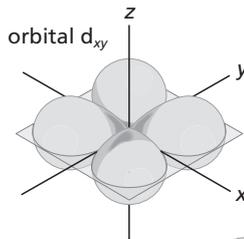


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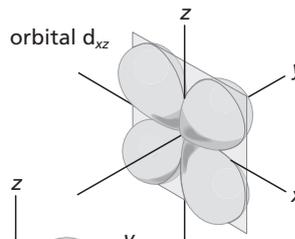


orbital p_z

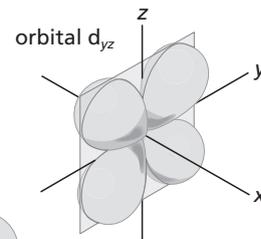
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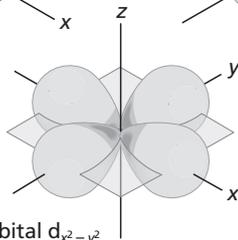
orbital d_{xy}



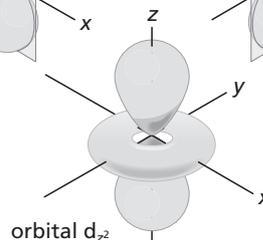
orbital d_{xz}



orbital d_{yz}

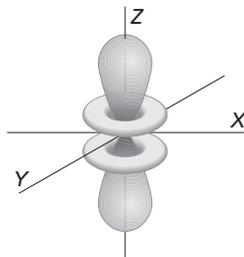


orbital $d_{x^2-y^2}$

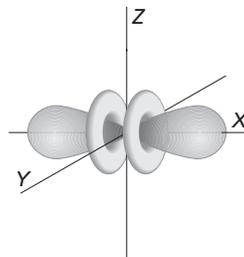


orbital d_{z^2}

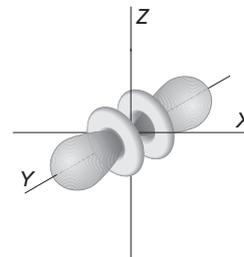
Orbitales d.



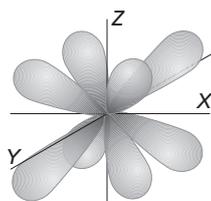
$f_{z^2 - \frac{3}{5} z r^2}$



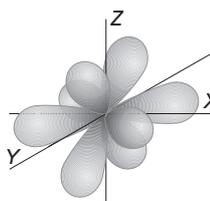
$f_{x^2 - \frac{3}{5} x r^2}$



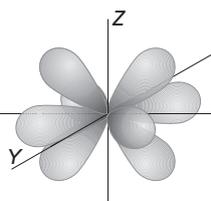
$f_{y^2 - \frac{3}{5} y r^2}$



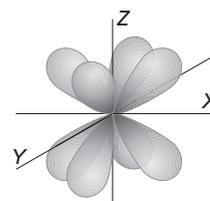
f_{xyz}



$f_{y(x^2-z^2)}$



$f_{x(z^2-y^2)}$



$f_{z(x^2-y^2)}$

Orbitales f.